HISTONE STOICHIOMETRY IN CHICKEN ERYTHROCYTE NUCLEI

Everline B. Wright and Donald E. Olins

University of Tennessee-Oak Ridge Graduate School of Biomedical Sciences
Biology Division, Oak Ridge National Laboratory
Oak Ridge, Tennessee 37830

Received January 30,1975

SUMMARY: In order to establish a quantitative relationship among the major histone components in chicken erythrocyte nuclei, an electrophoretic analysis was performed on purified histone fractions and on total histone extracts. The relative staining of the histones was calculated from quantitative densitometry of the amido black complexes. By comparison with the relative staining of the pure histone standards, we have calculated the molar ratios of the histones in the total histone extracts. The calculated molar ratios are not consistent with previous suggestions of pairs of all histones per repeating chromatin unit (the ν bodies).

There is much evidence that chromatin fibers consist of linear arrays of spheroid nucleoprotein particles (1-18), (denoted by us as ν bodies) (2). Several models have been proposed suggesting a specific molar stoichiometry of the individual histones in the repeating chromatin unit (the ν body) (2, 11, 16). To further the development of an adequate working model for the structure of chromatin, the relative molar ratios of the five major histone components in chicken erythrocyte nuclei were measured, and the results are reported in this paper.

METHODS: Chicken erythrocyte nuclei were isolated from the blood of young adult white Leghorn chickens as described previously (2, 19). Total histones were prepared from isolated nuclei by extraction with 0.25 M H₂SO₄, or with 5 M urea, 2 M NaCl as previously described (20). Pellets of the histones were dissolved in 0.01 N HCl, dialyzed exhaustively against 0.01 N acetic acid, lyophilized, and dissolved to a concentration of 1 mg/ml in 0.01 N HCl for gel electrophoresis. Individual histone fractions were prepared according to the method of Sanders and McCarty (21), and were observed to be electrophoretically pure. The purified histone fractions were stored frozen in water. Aliquots were removed for

amino acid analysis, and the remaining samples were quantitatively diluted in 0.9 N acetic acid containing 5% sucrose and 0.1% methyl green, for gel electrophoresis. Amino acid analyses were performed on a Beckman 120 C Amino Acid Analyzer (Beckman Instruments, Inc., Fullerton, California). From these analyses, the lysine content of each histone was determined. They were as follows: F2Al, 0.626 (µmol lysine/ml); F3, 0.327; F2A2, 0.779; F2B, 0.734; F1, 0.572; and F2C, 4.348. Using previously published data for the number of lysine residues per individual histone molecule, the molar concentrations of histone in the undiluted samples were calculated. Values for the number of lysine residues per histone molecule are as follows: F2Al, 11 (22); F3, 13 (23); F2A2, 14 (24); F2B, 20 (25); F1, 61 (26); and F2C, 51, assuming 24.1 moles % lysine and 212 amino acids per histone molecule (27).

Total histone extracts were analyzed on 15% acrylamide gels (0.5 X 24 cm), as described by Panyim and Chalkley (28, 29). Pre-electrophoresis was carried out at 100 volts overnight, and electrophoresis was performed at 200 volts for 16 hr. Purified histone fractions were analyzed on acid urea gels (0.5 X 8 cm), which were pre-electrophoresed overnight at 33 volts and electrophoresed at 100 volts for 4-6 hr. Gels were stained overnight in 0.1% amido black containing 20% ethanol and 7% acetic acid, and were destained by diffusion until the backgrounds were slightly blue. The gels were then removed from the destainer, placed in destaining solution containing several drops of 0.1% amido black, and allowed to equilibrate for several days until the backgrounds were uniform in color. Quantitative densitometric scans at 570 nm were recorded on a Gilford Model 2000 Spectrophotometer and analyzed as previously described (20).

All chemicals were reagent grade, and solutions were prepared with glass-distilled water.

RESULTS AND DISCUSSION: Total histone extracts and purified histone fractions were analyzed by gel electrophoresis, and the results of

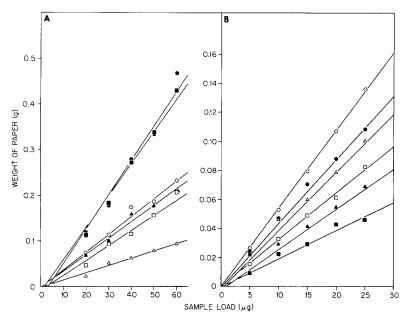


Figure 1: The results of quantitative densitometry on stained electrophoretic gels of total acid-extracted chicken erythrocyte histones (A) and on the purified chicken erythrocyte histones (B). The amount of stain absorbed at 570 nm by a histone peak is proportional to the weight of paper cut from a scan of that peak. The abscissa indicates the volume of sample loaded on the polyacrylamide gels prior to electrophoresis. Histones: F2Al (\square); F3 (0); F2A2 (\blacktriangle); F2B (\blacksquare); F1 (Δ); and F2C (\bullet).

quantitative densitometry are shown in Figure 1. These data indicate that there is a linear relationship between the area under each histone peak in the densitometric tracing (as measured by the paper weight) and the volume of sample loaded, thus confirming that: 1) Beer's Law is obeyed within the concentration range employed in this study and 2) the amount of stain absorbed per histone molecule remains constant as the quantity of histone is varied. From the slopes of least-squares lines fitted to the data of Figure 1A, the ratios of stain contained within the individual histone peaks (in total histone extracts) relative to the amount of stain absorbed by F2A1, were calculated; they are given in Table I (column 1). "Molar staining ratios" of the different histones were calculated from the slopes of a linear regression analysis of the data from Figure 1B, using the values of the molar concentrations of the

TABLE I									
	Relative	Histone	Content	in	Chicken	Erythrocyte	Nuclei		

Histone	1 Relative Staining (Total Histones)	2 Molar Staining (Pure Histones)	3 Molar Ratio (Total Histones)
F2A1	1.00	1.00	1.00
F3	0.99	1.39	0.71 ± 0.06*
F2A2	1.05	0.86	1.22 ± 0.10
F2B	2.08	0.92	2.26 ± 0.20
F1	0.51	1.97	0.26 ± 0.02
F2C	2.08	2.70	0.77 ± 0.05

^{*} Standard error values represent a minimum estimate, combining the standard errors of the slopes of the least-squares lines (Fig. 1A and 1B), neglecting errors in determining the histone concentrations and the moles of lysine per mole of histone. The total standard errors must be closer to $\pm 20\%$ of the calculated molar ratio.

pure histone fractions. The molar staining ratios are given relative to the staining of F2A1. As an example of the calculation of molar staining ratios, in one experiment the weight of paper from the F2C peak was 799.6 g/µmol of histone, that for F2Al was 296.1 g/µmol, yielding a molar staining ratio of 799.6/296.1, or 2.70/1 (see Table I, column 2). The third column in Table I contains values for the molar ratios of histones extracted from chicken erythrocyte nuclei and were calculated by dividing the values in column 1 by the corresponding values in column 2. Total histones extracted from purified chicken erythrocyte nuclei by treatment with 5 M urea, 2 M NaCl yielded values of relative staining that were very similar to those of acid-extracted histones. The standard errors calculated for the molar ratios of histones (Table I, column 3) represent minimum estimates based upon the errors of the least-squares regression analysis. They do not include the errors in the determination of histone concentrations by quantitative amino acid analysis or errors

in the assumed values of the moles of lysine per mole of histone. We can only guess that these combined errors would be larger than those shown, perhaps as much as ± 20% of the calculated molar ratios. With such total standard errors, and assuming that the molar ratios of the histones might be integral values, the data would be consistent with the following molar ratios of individual histone fractions in the total histone extract: F2A1, 1.0; F3, 1.0; F2A2, 1.0; F2B, 2.0; F1+F2C, 1.0. Such integral values must be taken with considerable caution, however, since the total errors in the analysis could not be adequately measured.

Significant differences in the relative molar staining ratios of the five major histone components of chicken erythrocyte nuclei (Figure 1B, Table I, column 2) were observed. Fambrough et al. (30) observed that the extinction coefficients of the amido black-histone complexes (dye bound per unit weight of protein) from pea chromatin were independent of the type of histone. We have recalculated our molar staining ratios into relative staining per unit weight of histone (assuming the observed molecular weights for the different histone fractions) and do not find equivalent extinction coefficients. We do not know the basis for this discrepancy between our work and the previous study, but we would point out that the previous authors (30) used polychromatic light in measuring the dye-absorbances of histone complexes. Recently, McMaster-Kaye and Kaye (31) obtained spectra of all of the amido black-histone complexes and concluded that there were differences among these spectra. Additionally, they found that the relative amount of stain absorbed by individual histones is a function of destaining. In the quantitative study reported here, gels of total histone extracts were destained and scanned identically to those of the purified histone fractions. Therefore, we feel that any preferential enhancement of the staining of one histone fraction over that of the others, caused by the choice of 570 nm as the wavelength for

TABLE II

Molar Ratios of Histones in Different Tissues

			F2A1	F3	F2A2	F2B	F1	F2C
Mammal								
Calf		thymus	1.00	0.77	1.15	1.46	0.54	
Calf		intestine	1.00	0.87	1.27	1.41	0.52	
Calf		spleen	1.00	0.36	0.72	1.26	0.50	
Calf		lung	1.00	0.73	1.41	1.91	0.59	
Calf		brain	1.00	1.13	1.14	1.53	0.59	
Calf		liver	1.00	0.53	0.79	1.71	0.40	
Calf		endometrium	1.00	0.86	1.14	1.78	0.39	
Calf		kidney	1.00	0.83	0.71	1.53	0.50	
Dog		spleen	1.00	1.18	1.23	2.30	1.09	
Dog		thymus	1.00	0.98	1.43	2.00	1.16	
Dog		1ung	1.00	0.78	0.74	1.90	0.35	
Opossi	ım	lung	1.00	2.38	1.37	2.32	0.76	
Opossi	ım	spleen	1.00	1.00	0.86	1.82	0.60	
Opossi	ım	thymus	1.00	1.54	2.32	4.11	0.63	
Opossi	ım	kidney	1.00	0.86	1.34	2.30	0.37	
Bird								
Hen		spleen	1.00	0.99	1.15	1.95	0.82	
Hen		blood	1.00	1.01	1.23	1.91	0.24	0.65
Hen		liver	1.00	1.01	1.19	1.92	0.80	
Hen		kidney	1.00	1.13	1.21	2.30	0.56	
Reptile								
Snapp:	ing turtle	kidney	1.00	1.99	3.21	4.07	1.47	
Snapp:	ing turtle	lung	1.00	0.50	0.74	1.33	0.25	
Snapp:	ing turtle	liver	1.00	2.08	2.49	5.51	1.49	
Snapp:	ing turtle	blood	1.00	1.19	1.97	2.65	0.89	
Amphibian								
Bullf:	cog	lung	1.00	1.12	1.37	1.96	0.86	
Bullf:	og	intestine	1.00	1.00	1.18	1.83	0.64	
Bullfi	cog	liver	1.00	0.90	1.08	1.15	0.44	
Bullf:	cog	blood	1.00	0.90	1.02	1.42	0.76	
Fish								
Carp		intestine	1.00	0.65	0.96	1.53	0.31	
Carp		liver	1.00	0.88	1.05	1.32	0.55	
Carp		blood	1.00	0.98	0.82	1.41	0.33	

scanning, or any undue loss of one histone component due to over-destaining, in the whole histone extract, has been compensated for by the similar treatment of the histone standards.

Our data on the molar ratios of nuclear histone extracts suggest that there is no simple quantitative relationship among these histones

unless assumptions of the total standard error are made (see above). However, this fact is not obvious if one considers only the percent staining of each individual histone band (Table I, column 1). Previous data on the relative staining of hen blood histone extracts stained with amido black and scanned at 600 nm (32), are in good agreement with the data presented here (scanned at 570 nm). When these relative staining data are combined with our measurements of the molar staining ratios, calculated molar histone ratios are obtained (Table II). In an attempt to determine if any pattern arises from the histones of other sources, we have calculated molar ratios from the extensive data of Panyim and Chalkley (29) and Panyim et al. (32), by using our molar staining ratio data. We realize that error is inherent in such calculations since: 1) molar staining ratios of histones of different species may vary and 2) the molar staining ratios reported in Table I were obtained at 570 nm, while the data of Panyim and Chalkley (29) and Panyim et al. (32) were measured at 600 nm. It is clear from these calculations (Table II) that there is no simple relationship among the five major types of histone in various chromatin species. Such data are not consistent with our previous suggestion of pairs of F2A1, F3, F2A2, F2B and F1 (or F2C) per vbody (2), or the later suggestion of Kornberg of pairs of F2A1, F3, F2B, F2A2 and one Fl per repeating unit (16). These data (Tables I and II) suggest at least two possibilities of histone stoichiometry per ν body: 1) If all ν bodies of any tissue type are alike, the histone molar ratios are not simple equimolar relationships and may differ in different tissues. 2) Alternatively, not all v bodies within a particular tissue type have identical histone stoichiometry.

ACKNOWLEDGEMENTS: We thank Drs. M. B. Senior, R. D. Carlson, F. D. Hamilton, F. H. Gaertner, and P. Mazur for helpful discussions, and Dr. F. C. Hartman for performing the amino acid analyses upon the pure

histone fractions. Dr. T. Mitchell kindly offered advice on the statistical treatment of the data. These studies were initiated by Ms. T. Whaley, an undergraduate summer trainee in the Carnegie Corporation program to train black students for careers in the Biomedical Sciences. Her enthusiasm and effort are gratefully acknowledged. This work was sponsored, in part, by the U.S. Atomic Energy Commission under contract with Union Carbide Corporation, and in part by NIGMS research grant (GM 19334) to D.E.O.

REFERENCES

- Olins, A. L., and Olins, D. E. (1973) J. Cell Biol. <u>59</u>, 252a.
 Olins, A. L., and Olins, D. E. (1974) Science <u>183</u>, <u>330</u>-332.
- 3. Olins, A. L., Senior, M. B., and Olins, D. E. (1974) J. Cell Biol. 63, 255a.
- 4. Senior, M. B., Olins, A. L., and Olins, D. E. (1975) Science 187, 173-175.
- Olins, A. L., Carlson, R. D., and Olins, D. E. (1975) J. Cell Biol., in press.
- 6. Woodcock, C. L. F. (1973) J. Cell Biol. 59, 368a.
- Woodcock, C. L. F., Mcguire, D. L., and Standfield, J. E. (1974) J. Cell Biol. 63, 377a.
- 8. Rill, R., and Van Holde, K. E. (1973) J. Biol. Chem. 248, 1080-1083.
- Sahasrabuddhe, C. G., and Van Holde, K. E. (1974) J. Biol. Chem. 249, 152-156.
- 10. Van Holde, K. E., Sahasrabuddhe, C. G., Shaw, B. R., Van Bruggen, E. F. J., and Arnberg, A. C. (1974) Biochem. Biophys. Res. Commun. 60, 1365-1370.
- 11. Van Holde, K. E., Sahasrabuddhe, C. G., and Shaw, B. R. (1974) Nucleic Acids Res. 1, 1579-1586.
- 12. Hewish, D. R., and Burgoyne, L. A. (1973) Biochem. Biophys. Res. Commun. <u>52</u>, 504-510.
- 13. Burgoyne, L. A., Hewish, D. R., and Mobbs, J. (1974) Biochem. J. <u>143</u>, 67-72.
- 14. Noll, M. (1974) Nature 251, 249-251.
- 15. Noll, M. (1974) Nucleic Acids Res. <u>1</u>, 1573-1578.
- 16. Kornberg, R. D. (1974) Science <u>184</u>, 865-868. 17. Axel, R., Melchior, W., Sollner-Webb, B., and Felsenfeld, G. (1974) Proc. Nat. Acad. Sci. U.S. 71, 4101-4105.
- 18. Langmore, J. P., and Wooley, J. C. (1974) J. Cell Biol. 63, 185a.
- 19. Olins, D. E., and Olins, A. L. (1972) J. Cell Biol. <u>57</u>, 715-736. 20. Olins, D. E., and Wright, E. B. (1973) J. Cell Biol. <u>59</u>, 304-317.
- 21. Sanders, L. A., and McCarty, K. S. (1972) Biochemistry 11, 4216-4222.
- 22. DeLange, R. J., Fambrough, D. M., Smith, E. L., and Bonner, J. (1969) J. Biol. Chem. <u>244</u>, 319-334.
- 23. DeLange, R. J., Hooper, J. A., and Smith, E. J. (1972) Proc. Nat.
- Acad. Sci. U.S. <u>69</u>, 882-884; (1972) J. Biol. Chem. <u>247</u>, 3589-3591. Yeoman, L. C., Olson, M. O. J., Sugano, N., Jordan, J. J., Taylor, C. W., Starbuck, W. C., and Busch, H. (1972) J. Biol. Chem. 247, 6018-6023.

- 25. Iwai, K., Ishikawa, K., and Hayashi, H. (1970) Nature <u>226</u>, 1056. 26. Bustin, M., and Cole, R. D. (1970) J. Biol. Chem. <u>245</u>, <u>1458-1466</u>. 27. Greenway, P. J., and Murray, K. (1971) Nature New <u>Biol</u>. <u>229</u>, 233-
- 28. Panyim, S., and Chalkley, R. (1969) Arch. Biochem. Biophys. 130, 337-346.
- 29. Panyim, S., and Chalkley, R. (1969) Biochemistry $\underline{8}$, 3972-3979. 30. Fambrough, D. M., Fujimura, F., and Bonner, J. (1968) Biochemistry
- 31. McMaster-Kaye, R., and Kaye, J. S. (1974) Anal. Biochem. 61, 120-132.
- 32. Panyim, S., Bilek, D., and Chalkley, R. (1971) J. Biol. Chem. 246, 4206-4215.